

Fill blank Ks

Suppose $20g = \text{C}_2\text{H}_6$ $5g = \text{C}_2\text{H}_4$

if $2 \text{C}_2\text{H}_6 + 3 \text{C}_2\text{H}_4 = \text{C}_2\text{H}_2\text{O}$

then $400g + \dots g = \dots g$

So $1200g$ of C_2H_6 react with $\dots g$ of C_2H_4 to get $\dots g$ of $\text{C}_2\text{H}_2\text{O}$

$100g$ of C_2H_6 react with $\dots g$ of C_2H_4 to get $\dots g$ of $\text{C}_2\text{H}_2\text{O}$

